



Short communication



Moringa Oleifera leaf extract mediated synthesis of reduced graphene oxide-vanadium pentoxide nanocomposite for enhanced specific capacitance in supercapacitors

T.N. Vinuth Raj^a, Priya A Hoskeri^{b,*}, Shanavaz Hamzad^c, M.S. Anantha^d, C.M. Joseph^b, H.B. Muralidhara^e, K. Yogesh Kumar^c, Fahad.A. Alharti^f, Byong-Hun Jeon^{g,**}, M.S. Raghu^{h,***}

^a Department of Physics, Faculty of Engineering and Technology, Jain University, Bangalore 562 112, India

^b Department of Physics, Dayananda Sagar College of Engineering, Kumaraswamy Layout, Bangalore 560 078, India

^c Department of Chemistry, Faculty of Engineering and Technology, Jain University, Bangalore 562 112, India

^d Department of Chemistry, HKBK College of Engineering, Bangalore 560045, India

^e Centre for Incubation, Innovation, Research & Consultancy, Jyothy Institute of Technology, Bangalore 560 082, India

^f Department of Chemistry, College of Science, King Saud University, Riyadh 11451, Saudi Arabia

^g Department of Earth Resources and Environmental Engineering, Hanyang University, 222, Wangsimni-ro, Seongdong-gu, Seoul 04763, Republic of Korea

^h Department of Chemistry, New Horizon College of Engineering, Outer Ring Road, Bangalore 560 103, India

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ABSTRACT

Moringa Oleifera leaf extract has been prepared and used as an environmental-friendly reducing agent. Reduced graphene oxide (RGO) has been decorated with vanadium pentoxide (V₂O₅) in presence of *Moringa Oleifera* leaf extract to generate a RGO/V₂O₅ nanocomposite. Analytical, spectroscopic, and electrochemical characterizations of synthesized V₂O₅, RGO, and RGO/V₂O₅ nanocomposite have been used to evaluate its morphology and structure. The RGO production and recombination with V₂O₅ are confirmed by the Characterization result. RGO, V₂O₅, and RGO/V₂O₅ nanocomposite have been evaluated for their electrochemical performances in supercapacitor applications by using various electrochemical techniques. The enhanced specific capacitance of 906 Fg⁻¹ has been accomplished for RGO/V₂O₅ nanocomposite compared to 146 and 535 Fg⁻¹ for V₂O₅ and RGO, respectively, at 2 mVs⁻¹ scan rate using cyclic voltammetry (CV) technique. Chronopotentiometry technique has also been evaluated and observed the specific capacitance of 380 Fg⁻¹ at a current density of 0.5 Ag⁻¹. RGO/V₂O₅ nanocomposite could be a choice of material in energy storage devices.

1. Introduction

Energy storage systems have garnered a lot of attention in recent decades, owing to increased demand for electronic devices, renewable energy cars and digital communications [1–5]. Because of improved power output, remarkable cycle life, low weight, ease of handling and flexibility supercapacitors have stimulated enormous interest in energy storage systems [6–8]. Supercapacitors could serve as next-generation power sources and probably plays a role in simple handy electronic devices like roll-up screens, tiny bio-medical devices, and wearable electronic gadgets due to their unique features [9–11]. However,

supercapacitors working voltage is restricted for potential implementation, leading to low energy density, which has hindered their prevalent adoption [12].

Supercapacitors have a wide range of uses, from electronic gadgets to electric cars to large-scale storage power grids, thanks to their unique electrochemical features. Supercapacitors are classified into two classes depending on their energy storage phenomena. Class-1 supercapacitors are electrical double-layer capacitors (EDLCs) which stores energy due to electrostatic adsorption/desorption at electrode/electrolyte interface [13,14]. Class-2 is pseudocapacitors that involve fast and reversible Faradaic redox reactions of the active materials [15]. In general,

* Corresponding author.

** Corresponding author.

*** Corresponding author.

E-mail addresses: priyahosin@gmail.com (P.A. Hoskeri), bhjeon@hanyang.ac.kr (B.-H. Jeon), raghu Hassan2009@gmail.com, dr.msraghu@newhorizonindia.edu (M.S. Raghu).

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